

E	Ethylene (C <sub>2</sub> H <sub>4</sub> , Ethene, R-1150)					
C.	AS: 74-85-1 UN: 1962				Ex Fla	
С	haracteristics				1	
	Flammable					
	Colourless gas with slight	odour				
	Colouriess gas with slight					
H	ealth Risks				100	
•	Asphyxiant at high conce	ntrations.				
Tr Al	ransport DR Class 2, 2F		DOT Class 2,1			
F	Product Description	Size (kg)	Material Number	Recommended Regulator		
E	Ethylene N3.0	16,7	541202-SH-C	W019120 or W019220		
E	Ethylene N3.5	18,5	541301-SO-C	W019120 or W019220		
E	Ethylene (N3.0) Purity		>99.9%			
N	Maximum Impurities (ppm)		$\begin{array}{c} CH_4, C_2H_6, C_3H_8\\ C_3H_6 \& \text{heavier}\\ C_2H_2\\ CO\\ CO_2\\ H\\ O_2^2\\ Total sulphur (as S)\\ H_2O \end{array}$	<900 <20 <20 <5 <10 <10 <10 <3 <5		
S	Stability Period (years)		5			
Ν	Material Code		541202-SH-C			
١	/alve		5/8" BSP LH female			
F	Pressure		91,1 bar			
_(	Cylinder		50 WC CrMo steel	cylinder		
Ν	Mass of Gas in Cylinder		16,7 kg			
	/olume of Gas @ 101.3 kPa		14,0 m₃			
\ (	absolute)					
\ (, F	absolute) Flammability in Air		3,1 - 32%			
\ ( F	absolute) Flammability in Air Applications		3,1 - 32% Polymer manufacture			

Standard Specifications

Maximum Impurities (ppm)

Ethylene N3.5 Purity (%)

<200

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	C <sub>2</sub> H <sub>2</sub>	<6
	CO	<3
	CO <sub>2</sub>	<10
	O <sub>2</sub>	<10
	H <sub>2</sub> O	<5
Stability Period (years)	5	
Material Code	541301-SO-C	
Valve	5/8" LH BSP Int	
Pressure	91,1 bar	
Cylinder	50 WC CrMo steel cylinder	
Mass of Gas in Cylinder	18,5 kg	
Volume of Gas @ 101,3 kPa (absolute)	15,5 m₃	
Flammability in Air	3,1 - 32%	
Applications	Polymer manufacture	
Precautions	Flammable gas under pressure	9

CH4, C2H6, C3H8 Polymer Grade

>99,95%

Physical Data	
Molecular Weight	28,054
Boiling Point at 1,013 bar [°C]	-103,68
Boiling Point at 14,5 psi [°F]	-154,60
Density at 1,013 bar, 20°C [kg/m₃]	1,173
Density at 1 atm, 70 ⁰F [lb/ft₃]	0,073
Vapour Pressure at 0°C [bar]	40,95
Vapour Pressure at 20°C [bar]	-
Vapour Pressure at 32°F [psi]	593,9
Vapour Pressure at 70°F [psi]	-
Flammability Range in Air [% volume]	2,7 - 36,0
Specific Volume at 1,013 bar, 20°C [m₃/kg]	0,859
Specific Volume at 1 atm, 70°F [ft <sub>3</sub> /lb]	13,7

Material Compatibility



## Source

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Special

- Ethylene is produced by passing ethyl alcohol vapours over dehydrating catalysts at 360 - 470°C.
- It may also be produced by the pyrolysis of ethane.
  - Cracking of petroleum is another source.

## Applications

- Ethylene is the starting material for several industrial syntheses. It is employed as an intermediate in the chemical industry and for the production of plastics.
- Ethylene is employed for the production of:
  - Acetaldehyde
  - Acetic acid
  - Chloroethane
  - Chloroethene (vinyl chloride)
  - Dichloroethane
  - 1,1-dichloroethene (vinylidene chloride)
  - Epoxyethane (ethylene oxide)
  - Ethanediol (ethylene glycol)
  - Ethanol
  - Ethoxyethane
  - Ethyl benzene
  - Phenylethene (styrene)
  - Polychloroethene (polyvinyl chloride)
  - Polythene

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- Propanoic acid
- Tetraethyl lead
- Trichloroethane.
- Ethylene is used as a component in calibration gases for the automotive, gas, oil as well as the chemical industries.
- Ethylene supplied in cylinders is used for controlled ripening of fruit, especially bananas. A concentration of a few ppm in the warehouse atmosphere is used. Because of flammability considerations, it is strongly recommended to use a mixture of ethene in nitrogen in this application. (see 'Ripegas')
- Ethylene has also been used in agriculture to promote crop growth. In this case the gas is injected directly into the soil.
- It is used as a refrigerant especially in the petrochemical